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THEORETICAL RESEARCH PROGRAM TO STUDY
CHEMICAL REACTIONS IN AOTV BOW SHOCK TUBES

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16. Abstract The main focus of this work was the development, implementation, and calibration of methods for performing molecular electronic structure calculations to high accuracy. These various methods were then applied to a number of chemical reactions and species of interest to NASA, notably in the area of combustion chemistry. Among the development work undertaken was a collaborative effort to develop a program to efficiently predict molecular structures and vibrational frequencies using energy derivatives. Another major development effort involved the design of new atomic basis sets for use in chemical studies: these sets were considerably more accurate than those previously in use. Much effort was also devoted to calibrating methods for computing accurate molecular wave functions, including the first reliable calibrations for realistic molecules using full CI results. A wide variety of application calculations were undertaken. One area of interest was the spectroscopy and thermochemistry of small molecules, including establishing small molecule binding energies to an accuracy rivaling, or even on occasion surpassing, the experiment. Such binding energies are essential input to modeling chemical reaction processes, such as combustion. Studies of large molecules and processes important in both hydrogen and hydrocarbon combustion chemistry were also carried out. Finally, some effort was devoted to the structure and spectroscopy of small metal clusters, with applications to materials science problems.			
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The main focus of the work carried out under Cooperative Agreement NCC2-371 was the development, implementation, and calibration of methods for performing molecular electronic structure calculations to a high degree of accuracy. These various methods were then applied to a number of chemical reactions and species of interest to NASA, notably in the area of combustion chemistry.

Among the development work undertaken was a collaborative effort (with several Scandinavian quantum chemists) to develop a program to efficiently predict molecular structures and vibrational frequencies using energy derivatives. This program exploits molecular symmetry using symmetry-adapted functions and can handle MCSCF wave functions. This combination of capabilities is unique. Another major development effort involved the design of new atomic basis sets for use in chemical studies. These "atomic natural orbital" sets were considerably more accurate than those previously in use, and they were utilized extensively by the entire NASA-Ames Computational Chemistry Branch in performing accurate quantum chemical calculations, as well as by outside researchers. Much effort was also devoted to calibrating methods for computing accurate molecular wave functions, including the first reliable calibrations for realistic molecules using full CI results. The full CI calculations were possible only on the NAS Facility CRAY-2 at the time (1985-1986), and they attracted considerable attention.

A wide variety of application calculations were undertaken. One area of interest was the spectroscopy and thermochemistry of small molecules, including establishing small molecule binding energies to an accuracy rivaling, or even on occasion surpassing, the experiment. Among the species studied was the radical CN, whose binding energy had been a long-standing controversy in the spectroscopy community, and the CH bond energy in acetylene, where two sets of conflicting experimental results were available. Our calculations resolved both these controversies. It may be noted that such binding energies are essential input to modeling chemical reaction processes, such as combustion. Among the more important

spectroscopic investigations were the rotation-vibration intensities (dipole moment function) of OH, a species important as a thermometric probe, and the electronic spectroscopy of diatomic aluminum halides, with relevance to radiation signatures. Studies of large molecules and processes important in both hydrogen and hydrocarbon combustion chemistry were also carried out, including binding energy studies of CH and NH bonds, and mechanistic studies of reactions important in prompt NO_x formation (CH + N₂) and in hydrocarbon combustion (acetylene pyrolysis). Finally, some effort was devoted to the structure and spectroscopy of small metal clusters, with applications to materials science problems, and to computing properties that influence nonlinear optical properties: polarizabilities and hyperpolarizabilities.

The following publications have resulted from this work:

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